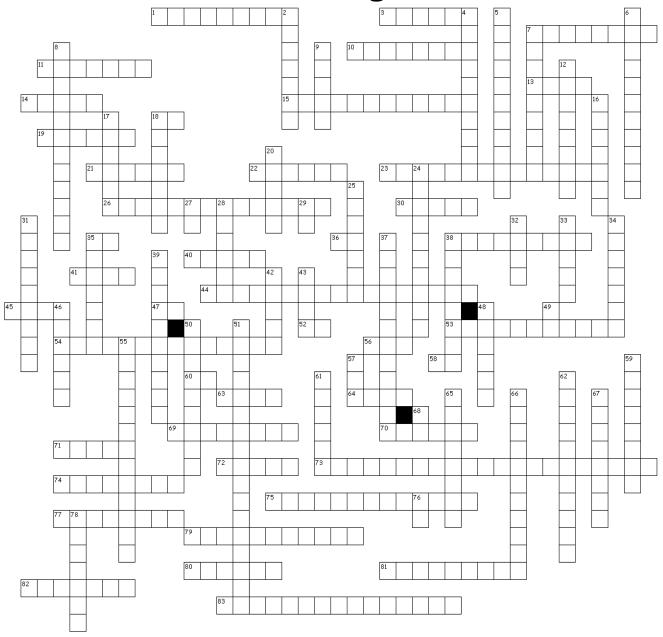
Bonding



Across

- particle smaller than the atom
- horizontal row in the periodic table
- division of an energy level
- two or more atoms of an element having 10. the same # protons, but different #neutrons
- particle found in the nucleus of an atom 11.
- 13.
- bond formed by transfer of electrons
- # protons
- a polar molecule
- positive ion
- positively charged subatomic particle
- shape with 3 bonds around central atom 26. force holding molecules to each other
- 30. negative ion
- 35. lead
- 36. osmium
- the study and investigation of the structure and properties of matter model for molecular shape based on
- repulsion of electron pairs 41. the force holding atoms together in a
- compound
- shape with 3 bonds + pair e 44.
- shape with 2 bonds, and 2 pair e 45.
- 47. hafnium
- antimony
- substance whose aqueous solution
- 54. force holding atoms together in molecules
- 56. Sn
- 58. platinum
- 60. gold
- smallest particle of an element possessing the properties of that element

- Ne
- equal sharing of electrons
- bond sharing two pairs of electrons
- element which tends to lose electrons in chemical reactions
- 73. description of the energy levels and sublevels for all the electrons in an atom
- element in group 18
- 75. pictorial arrangement of the elements based on their electron configurations
- unsymmetrical charge distribution
- the properties of the elements are a periodic function of their atomic numbers
- 80. vertical column in the periodic table
- the tendency of atoms to react to get 8 valence electrons
- neither positive nor negative
- exceptionally strong type of dipole 83. interaction

Down

- solid in which the particles are arranged in a regular, repeating pattern
- London forces
- 5. weak forces of attraction between molecules
- shape with 4 bonds around central atom 6
- has properties of metals and nonmetals 8.
- concerning the amounts of materials present
- 9. Rn
- two or more elements combined by chemical bonds
- 16. neutral subatomic particle
- 17. metal in group 1
- 18. element in group 17
- 20 vertical column in the periodic table energy required to remove an electron 24.

- mass per unit volume manganese
- shape with 2 bonds around central atom
- 29.
- 31. element which tends to gain electrons in chemical reactions
- 32 Zn
- group of three elements 33.
- the systematic investigation of nature
- substance produced as the result of a 35. chemical change
- 37. force holding metal atoms together, characterized by free electrons
- bond formed by sharing electrons
- the application of scientific principles for practical use
- 42 unequal sharing of electrons
- 43 the measure of the amount of matter
- 46. bond sharing three pairs of electrons
- 48. Ce
- 49. molybdenum
- 50. has properties of metals and nonmetals
- used to determine bond type
- metal in group 2
- 57. particle with electric charge 59
- negatively charged subatomic particle 61. bond sharing one pair of electrons
- 62. diagram of the location of the valence electrons
- neutral particles with two or more atoms held together with covalent bonds
- concerning the kinds of matter present
- substance that undergoes a chemical
- 68.
- space that can be occupied by 0, 1, or 2 electrons