

Chemical Names and Formulas

For the list on the left, name the compound. For the list on the right give the chemical formula that corresponds to the name. Remember to properly identify ions that have multiple charges.

1. KCl	_____	21. cesium chloride	_____
2. Na ₂ SO ₄	_____	22. ammonium nitrate	_____
3. BaF ₂	_____	23. sodium bromide	_____
4. Ca(OH) ₂	_____	24. calcium sulfide	_____
5. MgO	_____	25. magnesium phosphide	_____
6. CuS	_____	26. lithium nitride	_____
7. AgI	_____	27. potassium oxide	_____
8. Fe ₃ (PO ₄) ₂	_____	28. barium hydroxide	_____
9. NH ₄ Cl	_____	29. copper I nitrate	_____
10. Al(NO ₃) ₃	_____	30. iron III oxide	_____
11. CsNO ₂	_____	31. aluminum cyanide	_____
12. ZnCO ₃	_____	32. lead II sulfide	_____
13. KI	_____	33. strontium phosphate	_____
14. NH ₄ F	_____	34. nickel II sulfate	_____
15. CaBr ₂	_____	35. zinc hydroxide	_____
16. Na ₃ PO ₄	_____	36. magnesium iodide	_____
17. NaCl	_____	37. sodium nitrite	_____
18. FeCl ₃	_____	38. calcium sulfite	_____
19. AgClO ₃	_____	39. ammonium carbonate	_____
20. SrO	_____	40. iron III bromide	_____